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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/051,512	01/18/2002	Troy W. Francisco	H0002067 USA (4030)	1624
7590 05/20/2004			EXAMINER	
Colleen D. Szuch, Esq;			NGUYEN, NGOC YEN M	
Honeywell Law Dept. Honeywell International, Inc. 101 ColumbiaRoad, Building Meyer 5 Morristown, NJ 07962			ART UNIT	PAPER NUMBER
			1754	

DATE MAILED: 05/20/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

		Application No.	Applicant(s)
Office Action Summary		10/051,512	FRANCISCO ET AL.
		Examiner	Art Unit
		Ngoc-Yen M. Nguyen	1754
Period fo	<ul> <li>The MAILING DATE of this communication app or Reply</li> </ul>	ears on the cover sheet with the c	orrespondence address
THE - Exte after - If the - If NC - Failt Any	ORTENED STATUTORY PERIOD FOR REPLY MAILING DATE OF THIS COMMUNICATION. nsions of time may be available under the provisions of 37 CFR 1.13 SIX (6) MONTHS from the mailing date of this communication. e period for reply specified above is less than thirty (30) days, a reply period for reply is specified above, the maximum statutory period we are to reply within the set or extended period for reply will, by statute, reply received by the Office later than three months after the mailing ed patent term adjustment. See 37 CFR 1.704(b).	36(a). In no event, however, may a reply be time within the statutory minimum of thirty (30) days will apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	nely filed s will be considered timely. the mailing date of this communication. O (35 U.S.C. § 133).
Status			
	Responsive to communication(s) filed on <u>13 Au</u> This action is <b>FINAL</b> . 2b) This Since this application is in condition for allowar closed in accordance with the practice under E	action is non-final. nce except for formal matters, pro	
Disposit	ion of Claims		
5)⊠ 6)⊠	Claim(s) 1-12 and 14-19 is/are pending in the at 4a) Of the above claim(s) is/are withdraw Claim(s) 11,12,14,18 and 19 is/are allowed. Claim(s) 1-3,5-10,15 and 16 is/are rejected. Claim(s) 4, 17 is/are objected to. Claim(s) are subject to restriction and/or	vn from consideration.	
Applicat	ion Papers		
10)	The specification is objected to by the Examiner The drawing(s) filed on is/are: a) acce Applicant may not request that any objection to the o Replacement drawing sheet(s) including the correct The oath or declaration is objected to by the Ex	epted or b) objected to by the Edrawing(s) be held in abeyance. See on is required if the drawing(s) is obj	ected to. See 37 CFR 1.121(d).
Priority (	ınder 35 U.S.C. § 119		
12)∐ a)	Acknowledgment is made of a claim for foreign  All b) Some * c) None of:  1. Certified copies of the priority documents  2. Certified copies of the priority documents  3. Copies of the certified copies of the priorical application from the International Bureau  See the attached detailed Office action for a list of	s have been received. s have been received in Application ity documents have been receive (PCT Rule 17.2(a)).	on No d in this National Stage
Attachmen	t(s)		
2) 🔲 Notic 3) 🔲 Infori	e of References Cited (PTO-892) e of Draftsperson's Patent Drawing Review (PTO-948) mation Disclosure Statement(s) (PTO-1449 or PTO/SB/08) r No(s)/Mail Date	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal Pa 6) Other:	

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## **DETAILED ACTION**

Claims 11-12, 14, 18-19 are allowed.

Claims 4, 17 are objected to as being dependent upon a rejected base claim, but would be allowable if rewritten in independent form including all of the limitations of the base claim and any intervening claims.

The following is a statement of reasons for the indication of allowable subject matter: the prior art does not teach or suggest the use of an infrared analyzer equipped with a diamond-tipped probe to determine the concentration of free and bound water in an acid stream.

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-3, 5-10, 15-16 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 97/35187.

WO discloses a process for continuous production of hydrogen fluoride comprises (1) a step of reacting starting fluorspar with starting sulfuric acid, (2) a step of separating a crude reaction product as obtained in step (1) into (a) a low-boiling mixture composed predominantly of hydrogen

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fluoride and (b) a high-boiling mixture composed predominantly of unreacted sulfuric acid and containing small proportions of hydrogen fluoride and water,

(3) a step of purifying and isolating hydrogen fluoride from said low-boiling mixture (a), (4) a step of adding sulfuric anhydride to said high-boiling mixture (b) in a substantially equivalent amount with respect to the water to convert substantially all the water to sulfuric acid and returning it together with said unreacted sulfuric acid as sulfuric acid feed to said step (1), and (5) a step of adding sulfuric acid in a supplemental amount to provide the amount of sulfuric acid needed for reaction with starting fluorspar, wherein, in said step (4), the amount of water occurring in said high-boiling mixture (b) is determined by the method of the invention for determining the component concentration, particularly the concentration of water, of a ternary mixture (note page 4, lines 6-page 6, 31 and page 7, second full paragraph).

The process for determining the concentration of water is described in WO '187 as a method of determining the concentration of each component of a ternary mixture essentially consisting of sulfuric acid, hydrogen fluoride, and water, which comprises measuring at least one set of the three physical quantities, namely (1) temperature, (2) ultrasound propagation velocity, and (3) electrical conductivity or viscosity, of the ternary mixture and converting measured values to the concentrations of the respective components according to calibration curves representing the relationships of the concentrations of respective components of a ternary mixture composed of sulfuric acid, hydrogen

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fluoride, and water with the above-mentioned three physical quantities as separately constructed beforehand (note paragraph bridging pages 6-7).

WO '187 further discloses that the use of a computer is desired in order to conduct the treatment accurately and fast (note first paragraph on page 14).

The difference is WO '187 does not specifically disclose the use of a probe in at least a portion of the feed stream.

It would have been obvious to one of ordinary skill in the art at the time of the invention was made to use any known means in the art to performed the functions required in WO '187 in order to accurately estimate the amount of water, without a showing of criticality or unexpected results, the use such probe is not seen as a patentable difference since the probe is a known and commercially available means in the art.

Applicant's arguments filed August 13, 2003 have been fully considered but they are not persuasive.

Applicants argue that WO '187 fails to teach or suggest a method of regulating water content including the step of estimating the amount of bound and free water in a stream. This is fundamentally different that a method which measures just the total concentration of water in the stream.

It appears that in there are only two types of water in a mixture, bound (i.e., chemically bound) and free water. Therefore, there is no difference seen between the claimed "bound and free water" and the total concentration of water as disclosed in the

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applied reference. In the event that there is a difference, it should be noted that Applicants' claim 1 only require to "estimate", thus, the total concentration of water in the applied reference is a good estimate for the "bound and free water".

Applicants argue that WO '187 fails to tech or suggest any method for regulating and measuring water in a stream comprising sulfuric acid, hydrofluoric acid and fluorosulfonic acid, and water.

In WO '187, even though only sulfuric acid, hydrogen fluoride (i.e. hydrofluoric acid), and water are disclosed, not fluorosulfonic acid, however, since WO '187 discloses that the hydrogen fluoride is produced by reacting fluorspar and sulfuric, some fluorosulfonic acid may also be present in the mixture (note Applicants' specification, page 1, lines 22-29, which state that when fluorspar and acid are used to produce HF, an acid mixture is produced, which frequently comprises "a major proportion of sulfuric acid and minor proportions hydrofluoric acid and fluorosulfonic acid"). Furthermore, Applicants' claims do not require any particular amount for the fluorosulfonic acid, it can be present in the mixture in impurity amount.

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the

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shortened statutory period will expire on the date the advisory action is mailed, and any

extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of

the advisory action. In no event, however, will the statutory period for reply expire later

than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the

examiner should be directed to Ngoc-Yen M. Nguyen whose telephone number is (571)

272-1356. The examiner is currently on Part time schedule.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Stanley Silverman can be reached on (571) 272-1358. The fax phone

numbers for the organization where this application or proceeding is assigned are (703)

872-9306 for regular communications and (703) 872-9306 for After Final

communications.

Any inquiry of a general nature or relating to the status of this application or

proceeding should be directed to the receptionist whose telephone number is (703) 308-

0661.

Ngoc-Yen M. Nguyen Primary Examiner

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nmn

May 13, 2004